

Chemical Reactions



Name Class Date____ Match each of the following terms to its definition: Chemical change Conservation of mass Activation energy Chemical property Combustion Decomposition reaction Chemical reaction Catalyst - the minimum amount of energy needed to start a chemical reaction **2.** _____ - a substance that changes the rate of a chemical reaction by lowering the activation approximate are an example of actalysts new subs 4. **PREVIEW** into diffe Please Sign In or Sign Up to download the printable version of this worksheet 5. combine - a reaction between oxygen and another substance, usually compound + $O_2 \rightarrow product + heat$ an organic material; the burning of organic material 7. _____ - matter is not created or destroyed in a chemical reaction, the mass of the reactants must equal the mass of the products - a chemical reaction in which a compound is broken down into simpler compounds or elements



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Name Class Date Match each of the following terms to its definition: Chemical change Conservation of mass Activation energy Chemical property Combustion Chemical reaction Decomposition reaction Catalyst 1. activation energy the minimum amount of energy needed to start a chemical reaction 2. catalyst - a substance that changes the rate of a chemical reaction by activation lowering the activation approximation or an avample of actalysts 3. chem changed 4. chem **PREVIEW** change i Please Sign In or Sign Up to download the printable version of this worksheet 5. chem apart or **6. combustion** - a reaction between oxygen and another substance, usually compound $+ O_2 \longrightarrow product + heat$ an organic material; the burning of organic material 7. conservation of mass - matter is not created or destroyed in a chemical reaction; the mass of the reactants must equal the mass of the products **8. decomposition reaction** - a chemical reaction in which a compound is broken down into simpler compounds or elements