

Chemical Reactions



Name_____ Class_____ Date____ Match each of the following terms to its definition: Molecule Exothermic Decomposition reaction Conservation of mass Diatomic molecule Endothermic Physical change Double replacement reaction - matter is not created or destroyed in a chemical reaction; mass of products the mass of the reactants must equal the mass of the products 2. _____ - a chemical reaction in which a compound is broken down into simpler compounds or clements 3. hvdroger diatomic 4. **PREVIEW** ions; pro Please Sign In or Sign Up to download the printable version of this worksheet 5. surround - a chemical reaction that releases energy to the surroundings: burning wood is an exothermic reaction - the smallest unit of a substance made from a combination of elements - a change in which the appearance of a substance changes but its chemical properties remain unchanged; cutting paper is a physical change



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Name Class Date Match each of the following terms to its definition: Molecule Exothermic Decomposition reaction Conservation of mass Diatomic molecule Endothermic Physical change Double replacement reaction 1. conservation of mass - matter is not created or destroyed in a chemical mass of products reaction; the mass of the reactants must equal the mass of the products 2. decomposition reaction - a chemical reaction in which a compound is broken dawn into simpler compounds or elements 3. diator element: are diato 4. doubl + B D **PREVIEW** solution Please Sign In or Sign Up to download the printable version of this worksheet 5. endot surround **6. exothermic** - a chemical reaction that releases energy to the surroundings; burning wood is an exothermic reaction 7. molecule - the smallest unit of a substance made from a combination of elements \ **8. physical change** - a change in which the appearance of a substance changes but its chemical properties remain unchanged; cutting paper is a physical change