



Name _____ Class _____ Date _____

Match each of the following terms to its definition:

Boiling point

Amorphous solid

Colloid

Compound

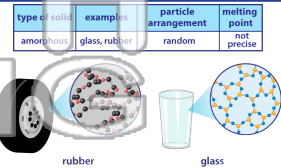
Concentration

Charles's law

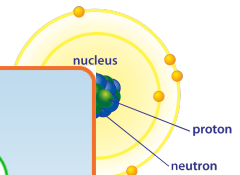
Atom

Boyle's law

1. _____ - a type of solid with randomly arranged particles and an imprecise melting point



2. _____ - the smallest particle or component of an element that still has the same properties of that element; composed of electrons, protons and neutrons



3. _____ a gas

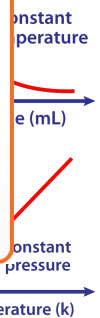


4. _____ constant

5. _____ pressure temperature

PREVIEW

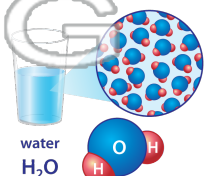
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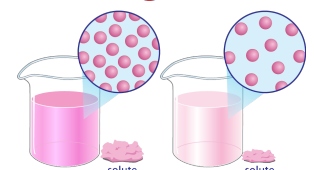
6. _____ - a mixture with evenly dispersed particles that are too small to see, but are large enough to scatter light and be filtered



7. _____ - a molecule made up of two or more different elements that are chemically combined



8. _____ - the amount of solute in a given amount of solvent





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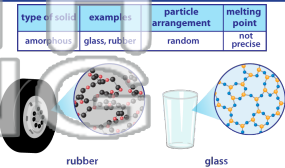
Concentration

Charles's law

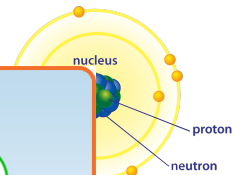
Atom

Boyle's law

1. **amorphous solid** - a type of solid with randomly arranged particles and an imprecise melting point



2. **atom** - the smallest particle or component of an element that still has the same properties of that element; composed of electrons, protons and neutrons

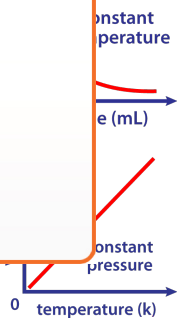


3. **boiling point** - the temperature at which a liquid changes to a gas



4. **Boyle's constant** - the constant in Boyle's law

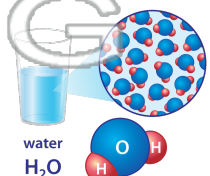
5. **Charles's law** - the relationship between the volume and temperature of a gas at constant pressure



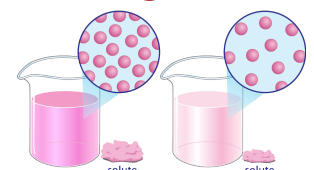
6. **colloid** - a mixture with evenly dispersed particles that are too small to see, but are large enough to scatter light and be filtered



7. **compound** - a molecule made up of two or more different elements that are chemically combined



8. **concentration** - the amount of solute in a given amount of solvent



PREVIEW

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