



Name \_\_\_\_\_ Class \_\_\_\_\_ Date \_\_\_\_\_

### Match each of the following terms to its definition:

Mixture

Surface tension

Molecule

Melting point

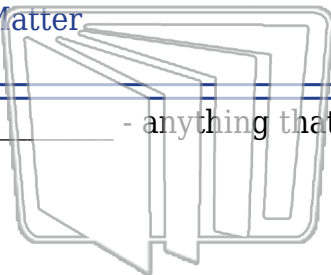
Matter

Solvent

Solute

Sublimation

1. \_\_\_\_\_ - anything that has mass and occupies space; a solid, liquid or gas



NEW PATH LEARNING



2. \_\_\_\_\_ - the temperature at which matter changes from a solid to a liquid; an ice cube's melting point is 0°Celsius



3. \_\_\_\_\_ materials

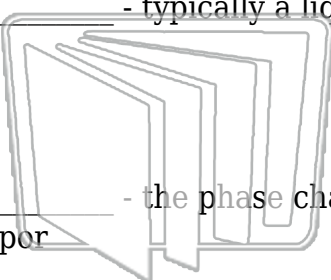


4. \_\_\_\_\_

## PREVIEW

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6. \_\_\_\_\_ - typically a liquid that dissolves a substance to form a solution

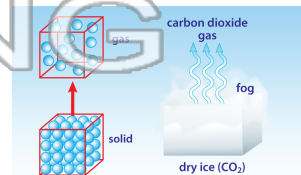


NEW PATH LEARNING

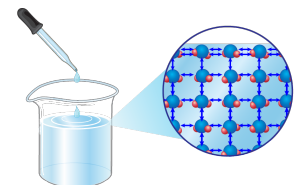


solute

7. \_\_\_\_\_ - the phase change that occurs when a solid goes directly into a vapor



8. \_\_\_\_\_ - a property of a liquid in which the molecules at the surface of a liquid have a strong attraction to neighboring molecules





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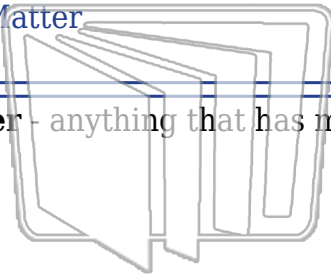
Matter

Solvent

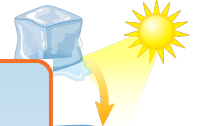
Solute

Sublimation

1. **matter** - anything that has mass and occupies space; a solid, liquid or gas



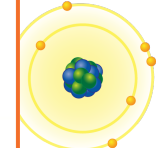
2. **melting point** - the temperature at which matter changes from a solid to a liquid; an ice cube's melting point is 0°Celsius



3. **mixture** substance



4. **molecule**

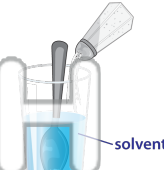
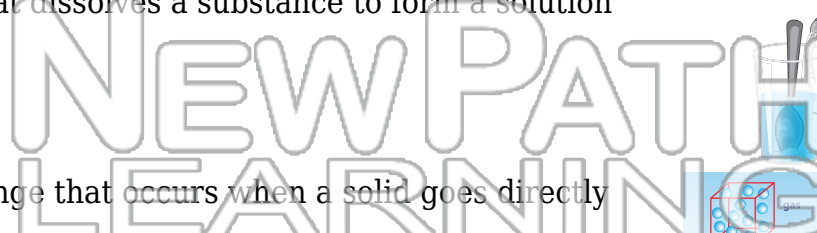
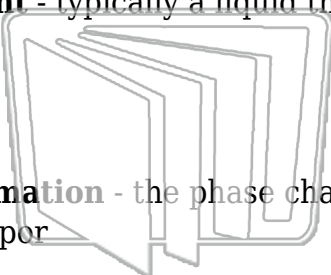


5. **solute**

**PREVIEW**  
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solute

6. **solvent** - typically a liquid that dissolves a substance to form a solution



7. **sublimation** - the phase change that occurs when a solid goes directly into a vapor



8. **surface tension** - a property of a liquid in which the molecules at the surface of a liquid have a strong attraction to neighboring molecules

