



Name _____ Class _____ Date _____

Match each of the following terms to its definition:

Single replacement reaction

Double replacement reaction

Endothermic reaction

Product

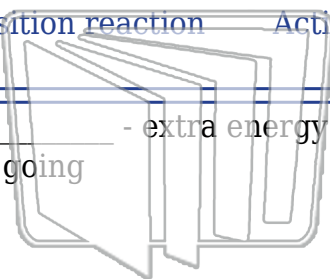
Decomposition reaction

Activation energy

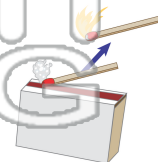
Reactant

Exothermic reaction

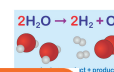
1. _____ - extra energy needed at the start of a reaction to get the reaction going



NEW PATH LEARNING



2. _____ - a chemical reaction in which a compound breaks down into individual elements



3. _____ parts of _____

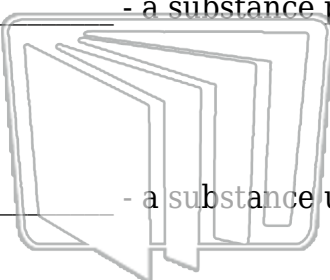


4. _____

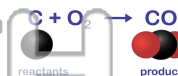
PREVIEW

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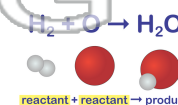
6. _____ - a substance produced during a chemical reaction



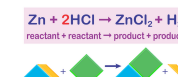
NEW PATH LEARNING



7. _____ - a substance used in a chemical reaction



8. _____ - a type of chemical reaction in which a single element replaces another element in a compound





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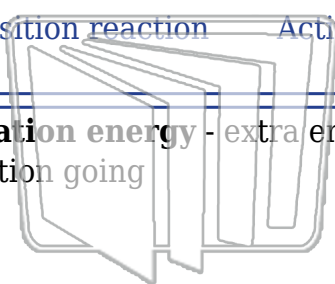
Decomposition reaction

Activation energy

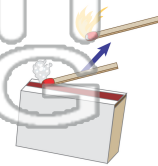
Reactant

Exothermic reaction

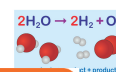
1. **activation energy** - extra energy needed at the start of a reaction to get the reaction going



NEW PATH LEARNING



2. **decomposition reaction** - a chemical reaction in which a compound breaks down into individual elements



3. **double replacement reaction** - a chemical reaction in which two elements or compounds exchange ions or atoms to form two new compounds



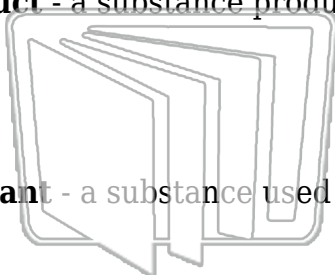
4. **endothermic reaction** - a chemical reaction in which energy is absorbed from the surroundings

PREVIEW

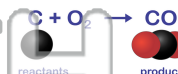
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5. **exothermic reaction** - a chemical reaction in which energy is released to the surroundings

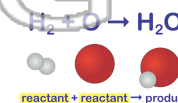
6. **product** - a substance produced during a chemical reaction



NEW PATH LEARNING



7. **reactant** - a substance used in a chemical reaction



8. **single replacement reaction** - a type of chemical reaction in which a single element replaces another element in a compound

