



Name _____ Class _____ Date _____

Match each of the following terms to its definition:

Double replacement reaction

Product

Exothermic reaction

Single replacement reaction

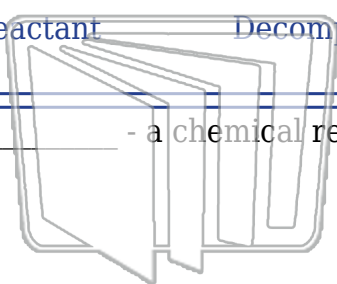
Reactant

Decomposition reaction

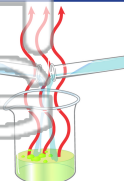
Synthesis reaction

Activation energy

1. _____ - a chemical reaction that releases energy

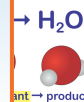


NEW PATH
LEARNING



2. _____ - a substance produced during a chemical reaction

3. _____



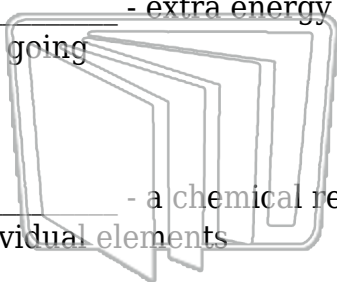
4. _____
replaces

PREVIEW

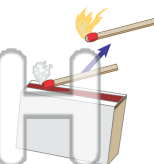
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5. _____
together
more cor

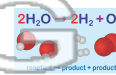
6. _____ - extra energy needed at the start of a reaction to get the reaction going



NEW PATH
LEARNING



7. _____ - a chemical reaction in which a compound breaks down into individual elements



8. _____ - a type of chemical reaction in which elements that are parts of compounds seem to exchange positions





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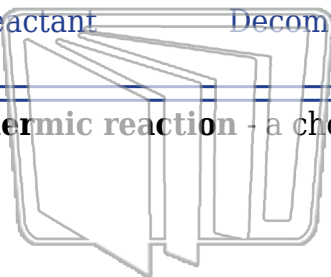
Reactant

Decomposition reaction

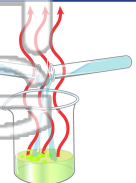
Synthesis reaction

Activation energy

1. **exothermic reaction** - a chemical reaction that releases energy

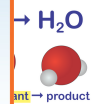


NEW PATH
LEARNING



2. **product** - a substance produced during a chemical reaction

3. **reactant**



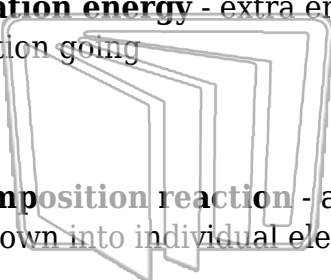
4. **single replacement reaction**

PREVIEW

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5. **synthesis reaction** - two or more substances put together to form a new substance

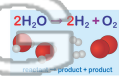
6. **activation energy** - extra energy needed at the start of a reaction to get the reaction going



NEW PATH
LEARNING



7. **decomposition reaction** - a chemical reaction in which a compound breaks down into individual elements



8. **double replacement reaction** - a type of chemical reaction in which elements that are parts of compounds seem to exchange positions

