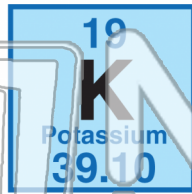




Name _____ Class _____ Date _____

1 The element **potassium** has _____.

- A 39 protons
- B 19 protons
- C 2 valence electrons
- D 55 neutrons



2 Identify the type of information.

- A atomic number
- B atomic mass
- C symbol
- D element name



3 Match the term with the correct definition.

- atomic mass unit** A. Different forms of the same element

4 Match the term with the correct definition.

- electron** A. Negatively charged particle found in _____



PREVIEW

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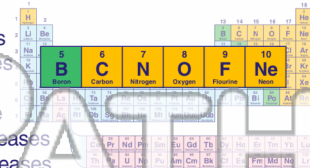
7 **noble gases?**

- A they do not react with other elements
- B they only react with metals
- C they only react with nonmetals
- D they react only with metalloids



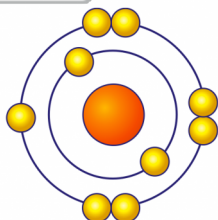
period in the periodic table?

- A increases
- B decreases
- C remains the same
- D both increases and decreases



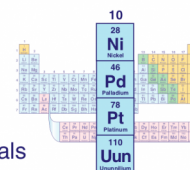
9 **Valence electrons** are always found in an atom's _____.

- A outermost orbit
- B innermost orbit
- C nucleus
- D compound



10 **Group 10** is just to the **left** of the groups containing the metalloids. Therefore, **all the atoms** in this group are considered to be _____.

- A nonmetals
- B noble gases
- C metals
- D semi-nonmetals

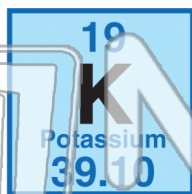




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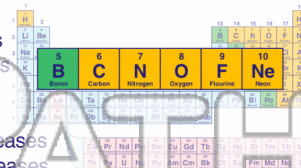
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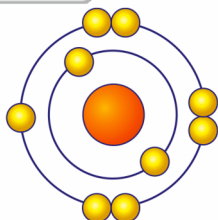
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