



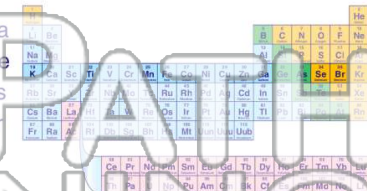
Name _____ Class _____ Date _____

1 Which trends are observed when the elements in **Period 3** on the Periodic Table are considered in order of **increasing atomic number**?

- A The atomic radius decreases, and the first ionization energy generally increases.
- B The atomic radius decreases, and the first ionization energy generally decreases.
- C The atomic radius increases, and the first ionization energy generally increases.
- D The atomic radius increases, and the first ionization energy generally decreases.

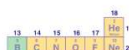
2 Which of the following **Period 4** elements has the **most metallic characteristics**?

- A Ca
- B Ge
- C As
- D Br



3 Which element is an **alkali metal**?

- A Na



4 In general, atoms of **transition elements** in **Period 5** are characterized by an incomplete

A 3s subshell

5

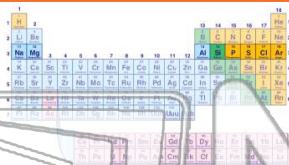


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- A protons
- B neutrons
- C valence electrons
- D occupied principal energy levels



- A have higher electronegativities
- B have lower first ionization energies
- C conduct electricity more readily
- D lose electrons more readily

9

Aqueous solutions of compounds containing element X are **blue**. Element X could be

- A carbon
- B copper
- C sodium
- D sulfur

10

Which element attains the structure of a **noble gas** when it becomes a **1+ ion**?

- A K
- B Ca
- C F
- D Ne





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