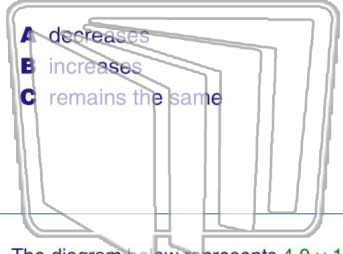




Name \_\_\_\_\_ Class \_\_\_\_\_ Date \_\_\_\_\_

1 As the **unbalanced force** applied to an object **increases**, the **time rate of change** of the object's momentum

- A decreases
- B increases
- C remains the same



2 A test booklet is sitting at rest on a desk. Compared to the magnitude of the force of the booklet on the desk, the **magnitude of the force of the desk on the booklet** is

- A less
- B greater
- C the same



# NEW PATH LEARNING

3 The diagram below represents  $4.0 \times 10^2$  kg satellite, S, in a circular orbit at an altitude of  $5.0 \times 10^6$  meters. The orbital speed of the satellite is  $5.0 \times 10^3$  meters per second and the radius of the Earth, R, is  $6.4 \times 10^6$  meters

4 The diagram below represents  $4.0 \times 10^2$  kg satellite, S, in a circular orbit at an altitude of  $5.0 \times 10^6$  meters. The orbital speed of the satellite is  $5.0 \times 10^3$  meters per second and the radius of the Earth, R, is  $6.4 \times 10^6$  meters



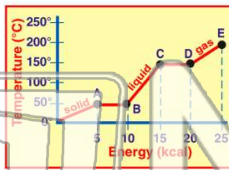
## PREVIEW

Please [Sign In](#) or [Sign Up](#) to download the printable version of this worksheet

7 pure substance and the heat energy added to the substance.

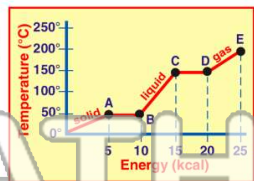
The **heat of fusion** of the substance is

- A 5.0 kcal/kg
- B 2.0 kcal/kg
- C 2.5 kcal/kg
- D 50 kcal/kg



the substance.  
The **specific heat of the substance in the liquid phase** is

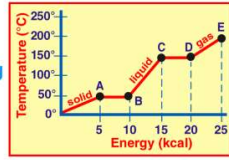
- A 0.025 kcal/kg-°C
- B 0.050 kcal/kg-°C
- C 20 kcal/kg-°C
- D 40 kcal/kg-°C



9 The graph below represents the relationship between the temperature of 2.0 kilograms of a pure substance and the heat energy added to the substance.

The **potential energy** of the molecules of the substance is **increasing** between points

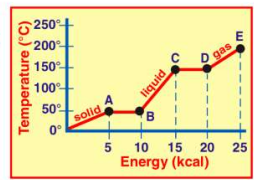
- A A and B, only
- B B and C, only
- C B and C, and D and E
- D A and B, and C and D



10 The graph below represents the relationship between the temperature of 2.0 kilograms of a pure substance and the heat energy added to the substance.

The **freezing point** of the substance is

- A 0° C
- B 50° C
- C 150° C
- D 200° C





Name \_\_\_\_\_ Class \_\_\_\_\_ Date \_\_\_\_\_

1 As the **unbalanced force** applied to an object **increases**, the **time rate of change** of the object's momentum



- A decreases
- B increases
- C remains the same

2 A test booklet is sitting at rest on a desk. Compared to the magnitude of the force of the booklet on the desk, the **magnitude of the force of the desk on the booklet** is



- A less
- B greater
- C the same

3 The diagram below represents  $4.0 \times 10^2$  kg satellite, S, in a circular orbit at an altitude of  $5.0 \times 10^6$  meters. The orbital speed of the satellite is  $5.0 \times 10^3$  meters per second and the radius of the Earth, R, is  $6.4 \times 10^6$  meters

4 The diagram below represents  $4.0 \times 10^2$  kg satellite, S, in a circular orbit at an altitude of  $5.0 \times 10^6$  meters. The orbital speed of the satellite is  $5.0 \times 10^3$  meters per second and the radius of the Earth, R, is  $6.4 \times 10^6$  meters



5

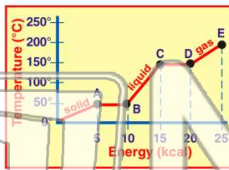
**PREVIEW**

Please [Sign In](#) or [Sign Up](#) to download the printable version of this worksheet

7 pure substance and the heat energy added to the substance.

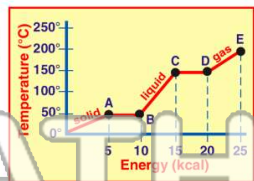
The **heat of fusion** of the substance is

- A 5.0 kcal/kg
- B 2.0 kcal/kg
- C 2.5 kcal/kg
- D 50 kcal/kg



the substance.  
The **specific heat** of the substance in the **liquid phase** is

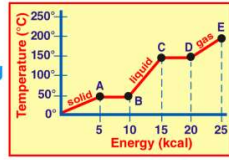
- A 0.025 kcal/kg-°C
- B 0.050 kcal/kg-°C
- C 20 kcal/kg-°C
- D 40 kcal/kg-°C



9 The graph below represents the relationship between the temperature of 2.0 kilograms of a pure substance and the heat energy added to the substance.

The **potential energy** of the molecules of the substance is **increasing** between points

- A A and B, only
- B B and C, only
- C B and C, and D and E
- D A and B, and C and D



10 The graph below represents the relationship between the temperature of 2.0 kilograms of a pure substance and the heat energy added to the substance.

The **freezing point** of the substance is

- A 0° C
- B 50° C
- C 150° C
- D 200° C

