

## **Properties of Matter**



Name Class Date Which property determines the direction Gas molecules at the same temperature are always assumed to have of the exchange of internal energy between two objects? A uniform velocity temperatu B uniform acceleration B specific h straight-line motion mass random motion density Equal masses of zinc and copper at room 3 As the volume of a fixed mass of an ideal temperature are placed in an oven that gas increases at constant temperature, supplies heat energy at a rate of 1 kilojoule the product of the pressure and the per minute. Compared to the time needed for 5 **PREVIEW** Please Sign In or Sign Up to download the printable version of this worksheet 7 the density of that gas A decrease by one A decreases B increase by one **B** increases C remain unchanged C remains the D decrease by tw 9 What is the minimum energy needed to ionize a hydrogen atom in the n = 2 **A**  $3.2 \times 10^{-19}$  C energy state? **B**  $4.5 \times 10^{-19}$  C A 13.6 eV **C**  $8.0 \times 10^{-19}$  C **B** 10.2 eV **D**  $9.6 \times 10^{-19}$  C C 3.40 eV **D** 1.89 eV



## Properties of Matter



Name Class Which property determines the direction Gas molecules at the same temperature are always assumed to have of the exchange of internal energy between two objects? A uniform velocity temperatu B uniform acceleration B specific h straight-line motion mass random motion density Equal masses of zinc and copper at room 3 As the volume of a fixed mass of an ideal temperature are placed in an oven that gas increases at constant temperature, supplies heat energy at a rate of 1 kilojoule the product of the pressure and the per minute. Compared to the time needed for 5 B **PREVIEW** Please Sign In or Sign Up to download the printable version of this worksheet 7 the density of that gas A decrease by one A decreases B B increase by one **B** increases C remain unchanged C remains the D decrease by two 9 What is the minimum energy needed to ionize a hydrogen atom in the n = 2 **A**  $3.2 \times 10^{-19}$  C energy state? **B**  $4.5 \times 10^{-19}$  C A 13.6 eV B **C**  $8.0 \times 10^{-19}$  C **B** 10.2 eV **D**  $9.6 \times 10^{-19}$  C C 3.40 eV **D** 1.89 eV